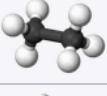
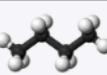
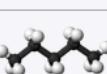
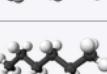
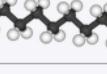
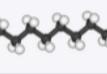
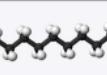
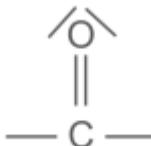
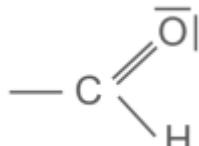
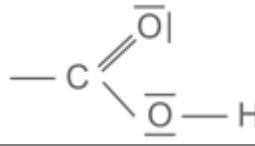
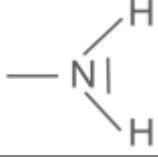
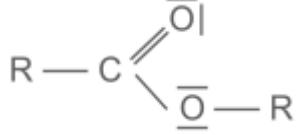
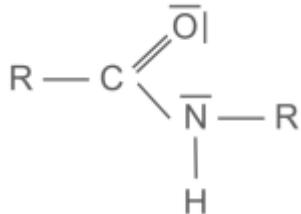


Die homologe Reihe der n-Alkane

Name	Summenformel	Molare Masse	Schmelzpunkt	Siedepunkt	Dichte	Kugel-Stab-Modell
Methan	CH ₄	16,04 g·mol ⁻¹	90,65 K	111,4 K	0,72 kg/m ³ gasförmig 0,42 g/cm ³ flüssig	
Ethan	C ₂ H ₆	30,07 g·mol ⁻¹	90 K	185 K	1,36 kg/m ³ (g) 0,54 g/cm ³ (fl.)	
Propan	C ₃ H ₈	44,10 g·mol ⁻¹	85 K	231 K	2,01 kg/m ³ (g) 0,58 g/cm ³ (fl.)	
n-Butan	C ₄ H ₁₀	58,12 g·mol ⁻¹	135 K	272,5 K	2,71 kg/m ³ (g) 0,60 g/cm ³ (fl.)	
n-Pentan	C ₅ H ₁₂	72,15 g·mol ⁻¹	144 K	309 K	0,626 g/cm ³	
n-Hexan	C ₆ H ₁₄	86,18 g·mol ⁻¹	178 K	342 K	0,659 g/cm ³	
n-Heptan	C ₇ H ₁₆	100,2 g·mol ⁻¹	182 K	371 K	0,684 g/cm ³	
n-Octan	C ₈ H ₁₈	114,2 g·mol ⁻¹	216 K	399 K	0,703 g/cm ³	
n-Nonan	C ₉ H ₂₀	128,3 g·mol ⁻¹	222 K	424 K	0,718 g/cm ³	
n-Decan	C ₁₀ H ₂₂	142,3 g·mol ⁻¹	243 K	447 K	0,73 g/cm ³	
n-Undecan	C ₁₁ H ₂₄	156,3 g·mol ⁻¹	248 K	469 K	0,74 g/cm ³	
n-Dodecan	C ₁₂ H ₂₆	170,3 g·mol ⁻¹	263 K	489 K	0,75 g/cm ³	

Stoffklassen und funktionelle Gruppen

Funktionelle Gruppe	Bezeichnung	Benennung in organischen Verbindungen	Verbundene Themen
$\begin{array}{c} \text{H} \\ \\ —\text{C}=\text{C}— \\ \\ \text{H} \end{array}$	Doppelbindung	-en	Additionsreaktion
$—\text{C}\equiv\text{C}—$	Dreifachbindung	-in	Additionsreaktion
$—\overline{\text{O}}—\text{H}$	Hydroxygruppe	hydroxy- / -ol	Alkohole Saccharide Ester

Funktionelle Gruppe	Bezeichnung	Benennung in organischen Verbindungen	Verbundene Themen
	Ketogruppe	-on	Ketone Carbonylverbindungen
	Aldehydgruppe	-al	Aldehyde Carbonylverbindungen
	Carboxygruppe	-carboxy	Carbonsäuren Ester
	Aminogruppe	-amin / -amino	Aminosäuren Peptidbindung Eiweiße
	Estergruppe	-ester	Ester Kondensationsreaktion
	Peptidbindung	-	Peptide Eiweiße Kondensationsreaktion

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